

Equally Efficient Interlayer Exciton Relaxation and Improved Absorption in Epitaxial and Nonepitaxial MoS₂/WS₂ Heterostructures

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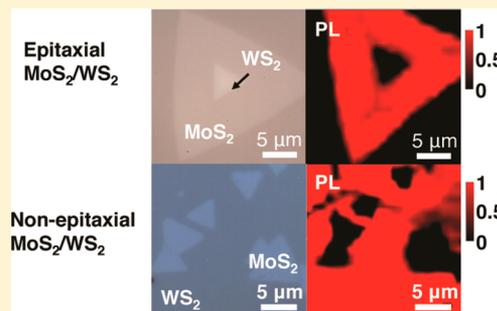
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S Supporting Information

ABSTRACT: Semiconductor heterostructures provide a powerful platform to engineer the dynamics of excitons for fundamental and applied interests. However, the functionality of conventional semiconductor heterostructures is often limited by inefficient charge transfer across interfaces due to the interfacial imperfection caused by lattice mismatch. Here we demonstrate that MoS₂/WS₂ heterostructures consisting of monolayer MoS₂ and WS₂ stacked in the vertical direction can enable equally efficient interlayer exciton relaxation regardless the epitaxy and orientation of the stacking. This is manifested by a similar 2 orders of magnitude decrease of photoluminescence intensity in both epitaxial and nonepitaxial MoS₂/WS₂ heterostructures. Both heterostructures also show similarly improved absorption beyond the simple superimposition of the absorptions of monolayer MoS₂ and WS₂. Our result indicates that 2D heterostructures bear significant implications for the development of photonic devices, in particular those requesting efficient exciton separation and strong light absorption, such as solar cells, photodetectors, modulators, and photocatalysts. It also suggests that the simple stacking of dissimilar 2D materials with random orientations is a viable strategy to fabricate complex functional 2D heterostructures, which would show similar optical functionality as the counterpart with perfect epitaxy.

KEYWORDS: Molybdenum disulfide, tungsten disulfide, van der Waals epitaxy, interlayer charge transfer, two-dimensional heterojunction



Engineering the dynamics of excitons, including generation, dissociation, transfer, and recombination, by semiconductor heterostructures bears tremendous significance for fundamental and applied interests.^{1,2} It stands as a major strategy for the development of all kinds of devices that involve photon-to-electron or electron-to-photon conversions, such as solar cells, LEDs, lasers, photodetectors, modulators, and photocatalysts. It also provides platforms with well-controlled excitons for the studies of fundamental physics. However, the capability of conventional semiconductor heterostructures to engineer excitons is often limited by the difficulty in developing high-quality interfaces for efficient interfacial charge transfer, a key step for the engineering of excitons. Typical heterostructures consist of two or more dissimilar semiconductor materials, and a nice match between the crystalline lattices of the semiconductor materials is required to yield high-quality interfaces. This requirement of lattice match imposes a fundamental constraint for the design of the conventional semiconductor heterostructures with increasing compositional

and structural complexity to provide sophisticated control of excitons.

Two-dimensional (2D) transition metal dichalcogenide (TMDC) materials in forms of a monolayer or fewlayer of atoms promise to enable a new type of semiconductor heterostructures. These materials present an atomic-scale semiconductor with bandgap in amplitude comparable to those of conventional group IV, III–V semiconductor materials. The heterostructures that consist of dissimilar 2D materials stacked in the vertical direction would provide capabilities to engineer excitons from a truly atomic level. Most importantly, unlike the conventional semiconductor heterostructures, which request lattice match to ensure high-quality interfaces, 2D heterostructures may have high quality interfaces regardless substantial lattice mismatch.^{3–10} This is because the interaction between 2D materials is van der Waals

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(vdW) forces and the weak interaction can relax the requirement of lattice match. Numerous works have recently demonstrated the fabrication of 2D heterostructures with the presence of lattice mismatch and the capability of the 2D heterostructures to efficiently engineer excitons.^{11–20} However, one very important question that has remained unanswered yet is how the excitonic properties of 2D heterostructures could depend on the epitaxy and orientation of the stacking. Knowledge of this question will provide useful guidance for the rational design of complex 2D heterostructures with desired exciton dynamics.

Here we have studied the excitonic properties of MoS₂/WS₂ heterostructures that consist of monolayer MoS₂ and WS₂ stacked either epitaxially or nonepitaxially in the vertical direction. Surprisingly, we demonstrate equally efficient interlayer relaxation of excitons in the heterostructures regardless the epitaxy and orientation of the stacking. This is manifested by a similar two-order magnitude decrease in the photoluminescence intensity of all of the heterostructures compared to that of separate monolayers. Additionally, both epitaxial and nonepitaxial MoS₂/WS₂ heterostructures show similarly improved absorption that is beyond the simple superimposition of the absorption of monolayer MoS₂ and WS₂, in particular for the incidence below the intrinsic bandgap of the monolayers. The nonepitaxial heterostructures are made by manually stacking single-crystalline monolayer MoS₂ and WS₂ pregrown separately with a chemical vapor deposition (CVD) process reported previously.^{10,21–25} The epitaxial heterostructures, which are single crystalline as well, are synthesized by a CVD process that we have developed with a mixture of MoO₃ and WO₃ as the precursors (see Methods and Supporting Information). Our result indicates that 2D heterostructures bear significant implications for the development of photonic devices, particularly those requesting efficient exciton separation and strong light absorption, such as solar cells,^{5,18–20} photodetectors, modulators, and photocatalysts. It also indicates that the simple stacking of dissimilar 2D materials with random stacking orientations may be a viable strategy to fabricate complex 2D heterostructures for the engineering of excitons.

We start the studies with epitaxial MoS₂/WS₂ heterostructures. Unlike nonepitaxial heterostructures, whose band structure is difficult to theoretically predict due to the difficulty in building up unit cells in theoretical models, the band structure of epitaxial MoS₂/WS₂ heterostructures has been well-studied using first principle techniques.^{11–14,26} This allows for synergistic studies from both experimental and theoretical sides to provide insights that are difficult to obtain from either perspective alone. The synthesized heterostructure consists of two concentric equilateral triangles in lateral size of tens of micrometers and well aligned in either the same or opposite directions (Figure 1a inset). Raman, STEM, and AFM characterizations indicate that the large triangle is single-crystalline monolayer MoS₂ while the small one single-crystalline monolayer WS₂, both continuous, smooth, and uniform (Figure 1a–b, and detailed characterizations seen in the Supporting Information). The STEM characterization also demonstrates that the MoS₂ and WS₂ monolayers, which have almost identical lattice constants,^{27,28} are epitaxially stacked together in an A–B staking mode along the vertical direction (Figure 1b and Figure S1–3).

We characterized the optical properties of the epitaxial MoS₂/WS₂ heterostructure at room temperatures. Figure 2a

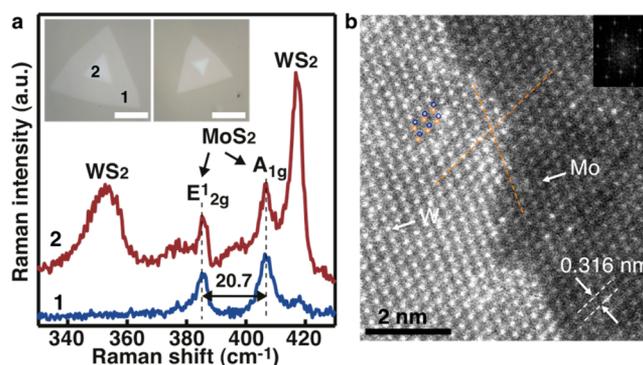


Figure 1. Characterizations of epitaxial MoS₂/WS₂ heterostructures. (a) Raman spectra collected from different areas (region 1 and 2) of the epitaxial heterostructure. The assignment for the Raman peaks is given as shown. Inset, optical images of two typical epitaxial MoS₂/WS₂ heterostructures with different relative orientations. The larger triangle is monolayer MoS₂, while the small one at the center is monolayer WS₂. Scale bar, 10 μm . (b) Scanning transmission electron microscope high-angle annular dark field image (STEM-HAADF) of the epitaxial heterostructure. The W and Mo atoms, which show different contrasts, are denoted in the figure. The lattice constant is measured in the figure as well. The dashed orange lines indicate the crystalline directions of the WS₂ layer. The circles in orange and blue represent W and Mo atoms, respectively, which are used to illustrate the offset of these atoms. Inset, the fast Fourier transformation pattern of the image.

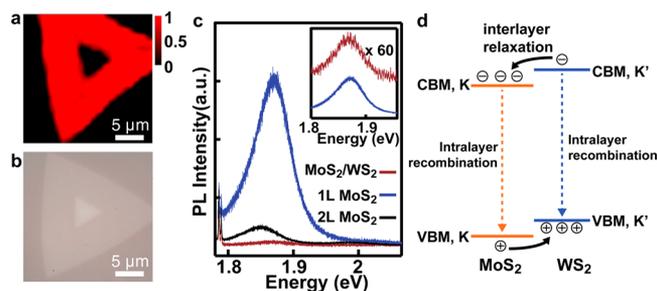


Figure 2. Low PL efficiency of epitaxial MoS₂/WS₂ heterostructures. (a) PL mapping of a typical epitaxial MoS₂/WS₂ heterostructure. (b) Optical image of the heterostructure mapped in (a). (c) Spectra PL collected from the monolayer (1L) MoS₂ area (red curve) and the MoS₂/WS₂ area (blue curve) of the heterostructure. The PL from a MoS₂ bilayer (2L) is also given (black curve). Inset, comparison of the PL from the MoS₂ area and the MoS₂/WS₂ area, where the PL from the MoS₂/WS₂ area scaled by a factor of 60 for visual convenience. (d) Schematic illustration for the bandstructure alignment of the heterostructure. The K point of MoS₂ coincides with the K' point of WS₂. The interlayer relaxations and intralayer recombination are illustrated.

shows the mapping of photoluminescence (PL) from a typical as-grown heterostructure whose optical image is given in Figure 2b. The structure consists of a small WS₂ monolayer in lateral size of $\sim 8 \mu\text{m}$ epitaxially stacked on the top of a big MoS₂ monolayer in lateral size of $\sim 25 \mu\text{m}$. We can immediately find that the PL from the edge region, which corresponds to monolayer MoS₂, is much stronger than that from the center where the MoS₂/WS₂ heterostructure is located. Representative PL spectra extracted from the mapping results are plotted in Figure 2c. The PL spectrum collected from the monolayer MoS₂ region (the big triangle) exhibits a strong peak at 1.87 eV, consistent with what was found for monolayer MoS₂ previously.²⁹ The PL collected from the MoS₂/WS₂ hetero-

structure region (the small triangle) shows a peak position similar to that of the MoS₂, but its intensity is smaller than that of the monolayer MoS₂ by 2 orders of magnitude (Figure 2c). To further illustrate the low PL efficiency of the heterostructure, the PL of bilayer MoS₂ collected under comparable conditions is given in Figure 2c as well (also see Figure S4). It has been well-known that bilayer MoS₂ shows weaker PL than monolayer MoS₂ because of the transition of the bandgap from direct to indirect.^{29–32} However, the PL intensity of the MoS₂/WS₂ heterostructure can be found even much weaker than that of bilayer MoS₂. We would like to point out two differences of our results from one recent study¹⁸ on similar epitaxial MoS₂/WS₂ heterostructures that was published during the review process of this work. First, the PL intensity of epitaxial MoS₂/WS₂ heterostructures in that study shows only three times smaller than that of monolayer MoS₂, instead of a 2 orders of magnitude decrease as we observed. Second, in that study, an additional PL peak at 1.4 eV was reported resulting from interlayer exciton transition, but we did not observe this PL peak in our materials even at a low temperature of 10 K (Figure S4). There are two possible reasons accounting for the differences. One could be the different synthetic processes, which might cause some differences in the resulting materials. Unlike that study, which used a mixture of element tungsten and tellurium as the precursor for WS₂, we used WO₃ instead. The other reason could be the difference in substrates, as sapphire substrates were used in our experiments while SiO₂/Si substrates were used in that study. It has been well-known that substrates could substantially affect the PL of 2D materials.³³

The established theoretical calculations for the bandstructure of epitaxial MoS₂/WS₂ heterostructures can provide useful insights into the fundamental physics underlying the observed low PL efficiency.^{11–14,26} Theoretical calculations have indicated that the band structure of epitaxial MoS₂/WS₂ heterostructures at the *K*-point in the Brillouin zone is approximately a simple superposition of the states of monolayer MoS₂ and WS₂.^{11,12,26} The MoS₂/WS₂ heterostructure essentially makes a type II heterojunction with the valence band maximum (VBM) completely localized to WS₂ and the conduction band minimum (CBM) to MoS₂.^{11,12,26} Indeed, we observed similar PL peak positions in the heterostructure and monolayer MoS₂ in experiments (Figure 2c), and this supports the theoretical prediction that the bandstructures at the *K* point may not change much after the heterostructuring. As a result, we conclude that the observed low PL of the heterostructure is due to the interlayer relaxation (dissociation) of excitons as illustrated in Figure 2d. The band structure offset between the MoS₂ and WS₂ monolayers can facilitate the separation of photoexcited charges, electrons to MoS₂ and holes to WS₂. This separation in different monolayers decreases the spatial overlap between the wave functions of electrons and holes, which may subsequently lead to the decrease in PL efficiency.³⁴ The observed low PL efficiency also indicates that the interlayer relaxation (dissociation) is very fast. As illustrated in Figure 2d, the interlayer relaxation of the photoexcited charges at band edges competes with another relaxation pathway, intralayer recombination. Given the simple superposition of the band structure as theoretically predicted,^{11,12,24} it is reasonable to assume that the intralayer recombination (at the *K* point) of the heterostructure is similar to that of standing-alone monolayers. The result that the PL of the heterostructure is 50–100 times weaker than that of monolayers MoS₂ implies that the interlayer relaxation process is 50–100 times faster

than the intralayer recombination in monolayer MoS₂. We can further roughly estimate the interlayer relaxation to be in a time scale of 10–100 fs as the intralayer recombination of excitons in monolayer MoS₂ is reported in scale of around 1–5 ps.^{35,36} This estimate is reasonably consistent with the result of another recent study¹⁷ that was published during the review process of this work, in which the interlayer transfer process in MoS₂/WS₂ heterostructures is experimentally measured to be within 50 fs. Note that the dramatic decrease in the PL efficiency of epitaxial MoS₂/WS₂ heterostructures we observed in experiments is actually different from what predicted in theory. The theoretical calculation did not fully recognize how efficient the interlayer relaxation could be and predicted substantial PL signal in the heterostructure due to the presence of direct transition at the *K* point.^{11,12}

Very surprisingly, to achieve the efficient interlayer relaxation of excitons does not require the heterostructure to be epitaxially stacked. We observed similarly efficient interlayer relaxation of excitons in nonepitaxial MoS₂/WS₂ heterostructures as well. To make the nonepitaxial MoS₂/WS₂ heterostructure, we first grew single-crystalline MoS₂ and WS₂ monolayers separately on sapphire substrates using the CVD processes reported previously^{23,37} and then transferred the monolayer MoS₂ onto the top of monolayer WS₂ using a unique surface energy-assisted transfer approach that we have recently developed³⁸ (see Methods and Figure S5). Different from the transfer techniques used in the previous works for the fabrication of heterostructures,^{17,23,26} which involved chemical etchants and would most likely cause damages and leave organic residues at the transferred materials, our surface energy-assisted approach relies on room temperature water droplet to transfer the monolayers and are proved able to better protect the quality of the transferred materials with no damage and organic residues left behind.³⁸ After the transfer, we mildly annealed the heterostructure at 200–250 °C for 10–30 min under an Ar flow to remove solvent or water residues. We have confirmed that both MoS₂ and WS₂ monolayers are very stable and this mild annealing process cannot cause any change in the quality and crystalline structures of the materials.

Figure 3a–b shows the result of PL mapping for typical nonepitaxial MoS₂/WS₂ heterostructure and the optical image of the heterostructures mapped. The heterostructures consist of numerous small single-crystalline monolayer MoS₂ in the lateral size of ~5 μm randomly stacked on top of a big single-crystalline WS₂ monolayer in size of ~50 μm (also see Figure S6). Similar to what we find with the epitaxial MoS₂/WS₂ heterostructure, the PL of all the nonepitaxial MoS₂/WS₂ heterostructures is 2 orders of magnitude weaker than that of monolayer MoS₂ or WS₂, regardless the relative orientation of the monolayers (Figure 3b–c), indicating the general presence of efficient interlayer exciton relaxation in all the heterostructures. The interlayer relaxation process is very sensitive to the surface quality of the heterostructures, which may affect the coupling between the MoS₂ and WS₂ monolayers. We find that the nonepitaxial MoS₂/WS₂ heterostructures, without being treated by the low temperature annealing (gray curve in Figure 3c), show a smaller decrease in PL intensity than the annealed one. The low-temperature annealing process may remove the residue of solvent and water molecules left between the two monolayers during the transfer process, which may subsequently facilitate the interlayer exciton relaxation. The independence of the efficient interlayer exciton relaxation in MoS₂/WS₂ heterostructures on the epitaxy and orientation of

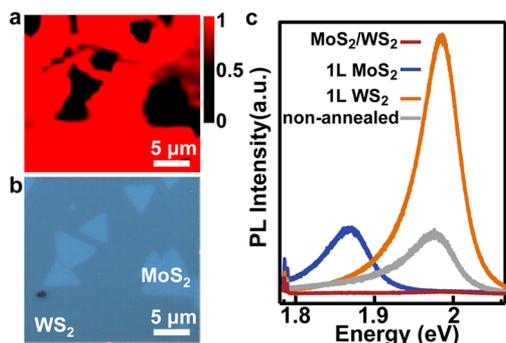


Figure 3. Low PL efficiency of nonepitaxial MoS_2/WS_2 heterostructures. (a) PL mapping of typical nonepitaxial MoS_2/WS_2 heterostructures. (b) Optical image of the heterostructure mapped in (a). It consists of multiple monolayer MoS_2 (small triangles) randomly distributed on top of a big monolayer WS_2 . This can be seen more clearly in the image given in Figure S6 that shows the edge of the monolayer WS_2 . (c) Spectra PL collected from the nonepitaxial MoS_2/WS_2 heterostructure (red curve), monolayer (1L) MoS_2 (blue curve), and monolayer (1L) WS_2 (brown curve). The PL from the nonepitaxial MoS_2/WS_2 heterostructure without being annealed is also given (gray curve).

the stacking suggests a strong electron–phonon coupling in 2D materials.³⁹ The electron–phonon coupling could be so strong that able to efficiently compensate whatever momentum mismatch for the charge transfer between the monolayers. A full-fledged study on the electron–phonon coupling is beyond the scope of this work.

To further understand our conclusion, we compare our results with what have been recently published during the review of this work.^{17,18,26} The 2 orders of magnitude decrease in PL intensity we observed is substantially greater than what reported by the other groups for nonepitaxial MoS_2/WS_2 heterostructures,^{17,18,26} which typically see a decrease of less than 3 times. The difference could be due to the different transferring processes used in the fabrication of the heterostructure. The unique surface energy-assisted transfer we used³⁸ can better protect the quality of the transferred monolayer with no organic residue and damage left than the approach used by the other groups.^{17,18,26} This could result in better coupling between the two monolayers involved and hence higher efficiency of interlayer exciton relaxation in the nonepitaxial heterostructure we made. Another possible reason could be related with the substrates. The two-order magnitude decreases in PL intensities we observed is from the nonepitaxial MoS_2/WS_2 heterostructures on sapphire substrates, while the nonepitaxial MoS_2/WS_2 heterostructures studied in some of the previous works^{18,26} were fabricated on SiO_2/Si substrates. We did observe a lesser decrease in PL at the nonepitaxial MoS_2/WS_2 heterostructures made on SiO_2/Si substrates (Figure S7). Nevertheless, the different decrease in the PL caused by using different substrates does not affect the generality of our conclusion. For instance, the decrease in PL that we observed at the nonepitaxial MoS_2/WS_2 heterostructures made on SiO_2/Si substrates (Figure S7) is very similar to the PL decrease recently observed at the epitaxial MoS_2/WS_2 heterostructures grown on SiO_2/Si substrates.¹⁸ This suggests that our conclusion for the independence of the interlayer exciton relaxation on the stacking epitaxy and orientation can be generally applied to the MoS_2/WS_2 heterostructures on other substrates, although using different substrates might lead to different absolute amplitudes of the decrease in PL

Whereas the PL is dramatically suppressed due to efficient interlayer exciton relaxation, MoS_2/WS_2 heterostructures show improved absorption that is beyond a simple superimposition of the absorptions of monolayer MoS_2 and WS_2 . The absorption improvement is particularly prominent for the incidence below the intrinsic band gap of the monolayers. The absorption spectra measured from epitaxial, annealed non-epitaxial, and nonannealed nonepitaxial MoS_2/WS_2 heterostructures are plotted in the upper panels of Figure 4 along with

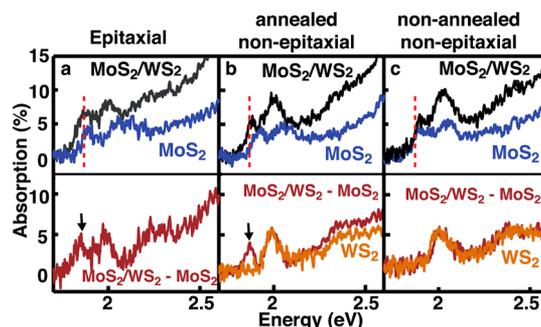


Figure 4. Improved absorption of MoS_2/WS_2 heterostructures. Absorption spectra collected from the MoS_2 area (blue curve) and the MoS_2/WS_2 area (black curve) for (a) epitaxial MoS_2/WS_2 heterostructures, (b) annealed nonepitaxial MoS_2/WS_2 heterostructures, and (c) nonepitaxial MoS_2/WS_2 heterostructure without being annealed. The dashed red lines indicate the boundary where the epitaxial and annealed nonepitaxial heterostructures show obvious higher absorption for the incidence to the left than the monolayer MoS_2 . The difference between the two absorption spectra ($\text{MoS}_2/\text{WS}_2 - \text{MoS}_2$) is given in the corresponding lower panel (red curve). For the nonepitaxial heterostructures, the absorption spectrum of monolayer WS_2 (brown curve) is also given in the lower panel as a reference. The black arrows point toward the peaks indicating the improved absorption of the heterostructures.

the absorption spectra of corresponding monolayer MoS_2 . We can find that the epitaxial and annealed nonepitaxial heterostructures exhibit substantially higher absorption efficiencies than monolayer MoS_2 for the incidence below the bandgap of MoS_2 , which is to the left of the dashed red lines (Figure 4a–b, upper panels). But similar absorption improvement for the sub-bandgap incidence cannot be found in the nonepitaxial heterostructure without being annealed (Figure 4c, upper panel). To further illustrate the relationship between the absorptions of the heterostructure and the monolayers involved, we subtract the absorption spectra of monolayer MoS_2 from those of MoS_2/WS_2 heterostructures. For the nonannealed nonepitaxial heterostructures, the spectrum resulted from the subtraction is identical to the absorption spectrum of monolayer WS_2 (Figure 4c, lower panel), indicating the absorption of the heterostructure is a simple superposition of separate monolayer MoS_2 and WS_2 . But for the epitaxial and annealed nonepitaxial heterostructures, the subtraction gives rise to a peak at 1.84 eV along with the features resulting from the absorption of monolayer WS_2 (Figure 4a–b, lower panels). Intuitively, this extra peak, which indicates improved absorption in the heterostructure for the low energy incidence, results from the red-shift of exciton peaks due to the reduction of dielectric screening in the heterostructure, similar to what observed in bilayer MoS_2 and bilayer WS_2 (Figure S8).⁴⁰

Our result indicates that 2D heterostructures present a useful platform for the engineering of excitons at the atomic level. For instance, it provides the capabilities to efficiently dissociate the excitons in 2D materials that would otherwise be difficult to separate and tend to radiatively recombine due to extraordinarily strong excitonic binding energy.^{41,42} The combination of efficient exciton dissociation and improved absorption make 2D heterostructures particularly useful for the absorption-based photonic devices, such as photovoltaics, solar fuels, photo-detectors, optical modulators, and photocatalysts. Additionally, the independence of the interlayer exciton relaxation on the stacking epitaxy and orientation clearly points out that the simple stacking of 2D materials in the vertical direction with random orientations is a viable strategy for the fabrication of functional 2D heterostructures. Complex 2D heterostructures fabricated by manually stacking dissimilar 2D materials with a random orientation may show equal optical functionality as the counterpart with perfect epitaxy.

Methods. Synthesis of Epitaxial MoS₂/WS₂ Heterostructures. The epitaxial heterostructures were synthesized by using a chemical vapor deposition process that we have developed by adapting what was previously reported for the synthesis of monolayer MoS₂ and WS₂.^{23,37} Briefly, the synthesis was performed in a tube furnace with sulfur (typically 1.0 g) and a mixture of MoO₃ and WO₃ (typically 80 mg with the weight percent of MoO₃ 1% and WO₃ 99%) as the precursors. The sulfur was placed at the upstream of the tube furnace and the mixed MoO₃/WO₃ at the center. Sapphire substrates were placed at the downstream in the tube. Other typical experimental conditions including a temperature of 950 °C and a flow of Ar gas in a rate of ~100 sccm. In a typical synthetic process, the temperature was ramped to 950 °C in 35 min and kept 950 °C for 2 h. After that, the whole system was cooled down to room temperature naturally.

Fabrication of Nonepitaxial MoS₂/WS₂ Heterostructures. The nonepitaxial heterostructures were made by manually stacking monolayer MoS₂ and WS₂ that were grown using the chemical vapor deposition process reported previously.^{23,37} The process is similar to what was used for the synthesis of heterostructures, but only either MoO₃ or WO₃ instead of the mixture was used for the synthesis of MoS₂ or WS₂. The typical temperatures used for the synthesis are 750 °C for MoS₂ and 900 °C for WS₂. Ar was used as the carrier gas for the synthesis of MoS₂ and forming gas (5% H₂ in Ar) for WS₂ with a flux rate of 100 sccm in both cases.

To make the nonepitaxial heterostructure, we first lifted off the synthesized MoS₂ from the growth substrate using a surface energy-assisted transfer process that we have recently developed.³⁸ Briefly, in a typical transfer process, 9 g of polystyrene (PS) with a molecular weight of 280 000 g/mol was dissolved in 100 mL of toluene, and then the PS solution was spin-coated (3500 rpm for 60 s) on the as-grown MoS₂ on sapphire substrates. This was followed by a baking at 80–90 °C for 15 min. A water droplet was then dropped on top of the sample. To facilitate the penetration of water molecules, we poked the PS layer with a sharp object from the edge. Once the PS layer was scratched from the edge, water molecules could penetrate through all the way under MoS₂, resulting the delamination of the PS-MoS₂ assembly. The water droplet was then removed away with paper towel. We could pick up the polymer/MoS₂ assembly with a tweezers and transferred it onto as-grown WS₂ monolayers. To ensure the uniformity of the transferred MoS₂, we baked the transferred PS-MoS₂ assembly

at 80 °C for 1 h and performed a final baking for 30 min at 150 °C. Finally, PS was removed by rinsing with toluene several times. After that, the stacked heterostructure was annealed at 200–250 °C for 10–30 min under an Ar environment.

Characterizations of MoS₂ Films. High-resolution STEM images were taken using the FEI Titan 80-300 probe aberration corrected and monochromated scanning transmission electron microscope (STEM) operated at 200 kV. In STEM mode, Z-contrast images were taken using a high-angle annular dark-field (HAADF) detector (Fischione Instrument), and elemental mapping was performed using the “Super X” energy dispersive spectrometric (EDS) system. We transferred the synthesized materials to TEM grids following a surface energy-assisted transfer approach that we have recently developed.³⁸ The convergence angle was set at 21 mrad, and probe current was about 110 pA, at which, we found that the beam damage on the MoS₂ sample can be controlled to the minimum within a reasonable period of time for imaging. The thickness and surface topology were measured using an atomic force microscope (AFM, Veeco Dimension-3000). Raman and photoluminescence (PL) measurements were carried out using Horiba Labram HR800 system using a 532 nm laser.

A home-built setup that consists of a confocal microscope (Nikon Eclipse C1) connected with a monochromator (SpectraPro, Princeton Instruments) and a detector (Pixis, Princeton Instruments) was used to perform the absorption measurement. In a typical measurement, we collected the white light transmitted through the sample using a 100× objective with a numerical aperture of 0.9 (Nikon). The light was from a halogen lamp and was broadly cast onto the samples with no focusing. We obtained the spectral absorption efficiency by normalized the transmitted light with I_1 and without the sample I_0 as $(I_0 - I_1)/I_0$. A focal plane aperture at the confocal scanning head installed with the microscope allows us to define the sample area to be measured with a spatial resolution of 300 nm.

■ ASSOCIATED CONTENT

📄 Supporting Information

(1) Detailed experimental results on the synthesis and compositional and structural characterizations of the synthesized epitaxial MoS₂/WS₂ heterostructures; (2) more Raman and optical measurements for bilayer MoS₂, bilayer WS₂, and MoS₂/WS₂ heterostructures. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Author Contributions

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Notes

The authors declare no competing financial interest.

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REFERENCES

- (1) Weisbuch, G.; Vinter, B. *Quantum Semiconductor Structures: Fundamentals and Applications*; Academic Press: San Diego, 1991.
- (2) Piprek, J. *Semiconductor Optoelectronic Devices*; Academic Press: San Diego, 2003.
- (3) Gao, G. H.; Gao, W.; Cannuccia, E.; Taha-Tijerina, J.; Balicas, L.; Mathkar, A.; Narayanan, T. N.; Liu, Z.; Gupta, B. K.; Peng, J.; Yin, Y. S.; Rubio, A.; Ajayan, P. M. *Nano Lett.* **2012**, *12*, 3518–3525.
- (4) Geim, A. K.; Grigorieva, I. V. *Nature* **2013**, *499*, 419–425.
- (5) Britnell, L.; Ribeiro, R. M.; Eckmann, A.; Jalil, R.; Belle, B. D.; Mishchenko, A.; Kim, Y. J.; Gorbachev, R. V.; Georgiou, T.; Morozov, S. V.; Grigorenko, A. N.; Geim, A. K.; Casiraghi, C.; Neto, A. H. C.; Novoselov, K. S. *Science* **2013**, *340*, 1311–1314.
- (6) Shi, Y. M.; Zhou, W.; Lu, A. Y.; Fang, W. J.; Lee, Y. H.; Hsu, A. L.; Kim, S. M.; Kim, K. K.; Yang, H. Y.; Li, L. J.; Idrobo, J. C.; Kong, J. *Nano Lett.* **2012**, *12*, 2784–2791.
- (7) Georgiou, T.; Jalil, R.; Belle, B. D.; Britnell, L.; Gorbachev, R. V.; Morozov, S. V.; Kim, Y.-J.; Gholinia, A.; Haigh, S. J.; Makarovskiy, O.; Eaves, L.; Ponomarenko, L. A.; Geim, A. K.; Novoselov, K. S.; Mishchenko, A. *Nat. Nanotechnol.* **2013**, *8*, 100–103.
- (8) Yu, W. J.; Li, Z.; Zhou, H.; Chen, Y.; Wang, Y.; Huang, Y.; Duan, X. *Nat. Mater.* **2013**, *12*, 246–252.
- (9) Zhang, W.; Chuu, C.-P.; Huang, J.-K.; Chen, C. H.; Tsai, M.-L.; Chang, Y.-H.; Liang, C.-T.; He, J.-H.; Chou, M.-Y.; Li, L.-J. *Sci. Rep.* **2013**, *4*, 3826.
- (10) Ji, Q. Q.; Zhang, Y. F.; Gao, T.; Zhang, Y.; Ma, D. L.; Liu, M. X.; Chen, Y. B.; Qiao, X. F.; Tan, P. H.; Kan, M.; Feng, J.; Sun, Q.; Liu, Z. F. *Nano Lett.* **2013**, *13*, 3870–3877.
- (11) Komsa, H.-P.; Krasheninnikov, A. V. *Phys. Rev. B* **2013**, *88*, 085318.
- (12) Kořmider, K.; Fernández-Rossier, J. *Phys. Rev. B* **2013**, *87*, 075451.
- (13) Terrones, H.; López-Urías, F.; Terrones, M. *Sci. Rep.* **2013**, *3*, 1549.
- (14) Kang, J.; Tongay, S.; Zhou, J.; Li, J. B.; Wu, J. Q. *Appl. Phys. Lett.* **2013**, *102*.
- (15) Fang, H.; Battaglia, C.; Carraro, C.; Nemsak, S.; Ozdol, B.; Kang, J. S.; Bechtel, H. A.; Desai, S. B.; Kronast, F.; Unal, A. A.; Conti, G.; Conlon, C.; Palsson, G. K.; Martin, M. C.; Minor, A. M.; Fadley, C. S.; Yablonovitch, E.; Maboudian, R.; Javey, A. *Proc. Natl. Acad. Sci. U.S.A.* **2014**, *111*, 6198–6202.
- (16) Rivera, P.; Schaibley, J. R.; Jones, A. M.; Ross, J. S.; Wu, S.; Aivazian, G.; Klement, P.; Ghimire, N. J.; Yan, J.; Mandrus, D. G.; Yao, W.; Xu, X. *arXiv:1403.4985* **2014**.
- (17) Hong, X.; Kim, J.; Shi, S.-F.; Zhang, Y.; Jin, C.; Sun, Y.; Tongay, S.; Wu, J.; Zhang, Y.; Wang, F. *Nat. Nanotechnol.* **2014**, *9*, 682–686.
- (18) Gong, Y.; Lin, J.; Wang, X.; Shi, G.; Lei, S.; Lin, Z.; Zou, X.; Ye, G.; Vajtai, R.; Yakobson, B. I.; Terrones, H.; Terrones, M.; Tay, B. K.; Lou, J.; Pantelides, S. T.; Liu, Z.; Zhou, W.; Ajayan, P. M. *Nat. Mater.* **2014**, *13*, 1135–1142.
- (19) Pospischil, A.; Furchi, M. M.; Mueller, T. *Nat. Nanotechnol.* **2014**, *9*, 257–261.
- (20) Lee, C.-H.; Lee, G.-H.; Zande, A. M. v. d.; Chen, W.; Li, Y.; Han, M.; Cui, X.; Arefe, G.; Nuckolls, C.; Heinz, T. F.; Guo, J.; Hone, J.; Kim, P. *Nat. Nanotechnol.* **2014**, *9*, 676–681.
- (21) Najmaei, S.; Liu, Z.; Zhou, W.; Zou, X. L.; Shi, G.; Lei, S. D.; Yakobson, B. I.; Idrobo, J. C.; Ajayan, P. M.; Lou, J. *Nat. Mater.* **2013**, *12*, 754–759.
- (22) van der Zande, A. M.; Huang, P. Y.; Chenet, D. A.; Berkelbach, T. C.; You, Y. M.; Lee, G. H.; Heinz, T. F.; Reichman, D. R.; Muller, D. A.; Hone, J. C. *Nat. Mater.* **2013**, *12*, 554–561.
- (23) Lee, Y. H.; Zhang, X. Q.; Zhang, W.; Chang, M. T.; Lin, C. T.; Chang, K. D.; Yu, Y. C.; Wang, J. T. W.; Chang, C. S.; Li, L. J. *Adv. Mater.* **2012**, *24*, 2320–2325.
- (24) Butler, S. Z.; Hollen, S. M.; Cao, L.; Cui, Y.; Gupta, J. A.; Gutiérrez, H. R.; Heinz, T. F.; Hong, S. S.; Huang, J.; Ismach, A. F.; Johnston-Halperin, E.; Kuno, M.; Plashnitsa, V. V.; Robinson, R. D.; Ruoff, R. S.; Salahuddin, S.; Shan, J.; Shi, L.; Spencer, M. G.; Terrones, M.; Windl, W.; Goldberger, J. E. *ACS Nano* **2013**, *7*, 2898–2926.
- (25) Lee, Y.-H.; Yu, L.; Wang, H.; Fang, W.; Ling, X.; Shi, Y.; Lin, C.-T.; Huang, J.-K.; Chang, M.-T.; Chang, C.-S. *Nano Lett.* **2013**, *13*, 1852–1857.
- (26) Tongay, S.; Fan, W.; Kang, J.; Park, J.; Koldemir, U.; Suh, J.; Narang, D. S.; Liu, K.; Ji, J.; Li, J.; Sinclair, R.; Wu, J. *Nano Lett.* **2014**, *14*, 3185–3190.
- (27) Schonfeld, B.; Huang, J.; Moss, S. *Acta Crystallogr., Sect. B: Struct. Sci.* **1983**, *39*, 404–407.
- (28) Schutte, W.; De Boer, J.; Jellinek, F. J. *Solid State Chem.* **1987**, *70*, 207–209.
- (29) Yu, Y. F.; Li, C.; Liu, Y.; Su, L. Q.; Zhang, Y.; Cao, L. Y. *Sci. Rep.* **2013**, *3*, 1866.
- (30) Mak, K. F.; Lee, C.; Hone, J.; Shan, J.; Heinz, T. F. *Phys. Rev. Lett.* **2010**, *105*, 136805.
- (31) Splendiani, A.; Sun, L.; Zhang, Y. B.; Li, T. S.; Kim, J.; Chim, C. Y.; Galli, G.; Wang, F. *Nano Lett.* **2010**, *10*, 1271–1275.
- (32) Cheiwchanchamnangij, T.; Lambrecht, W. R. L. *Phys. Rev. B* **2012**, *85*, 205302.
- (33) Buscema, M.; Steele, G. A.; Zant, H. S. J. v. d.; Castellanos-Gomez, A. *Nano Res.* **2014**, *7*, 561–571.
- (34) Miller, D. A. B. *Quantum Mechanics for Scientists and Engineers*; Cambridge University Press: New York, 2008.
- (35) Korn, T.; Heydrich, S.; Hirmer, M.; Schmutzler, J.; Schuller, C. *Appl. Phys. Lett.* **2011**, *99*, 102109.
- (36) Mai, C.; Barrette, A.; Yu, Y. F.; Semenov, Y. G.; Kim, K. W.; Cao, L. Y.; Gundogdu, K. *Nano Lett.* **2014**, *14*, 202–206.
- (37) Gutierrez, H. R.; Perea-Lopez, N.; Elias, A. L.; Berkdemir, A.; Wang, B.; Lv, R.; Lopez-Urías, F.; Crespi, V. H.; Terrones, H.; Terrones, M. *Nano Lett.* **2013**, *13*, 3447–3454.
- (38) Gurarslan, A.; Yu, Y.; Su, L.; Yu, Y.; Suarez, F.; Yao, S.; Zhu, Y.; Ozturk, M.; Zhang, Y.; Cao, L. *ACS Nano* **2014**, *8*, 11522–11528.
- (39) Fivaz, R.; Mooser, E. *Phys. Rev.* **1964**, *136*, A833.
- (40) Consadori, F.; Frindt, R. *Phys. Rev. B* **1970**, *2*, 4893.
- (41) Mak, K. F.; He, K. L.; Lee, C.; Lee, G. H.; Hone, J.; Heinz, T. F.; Shan, J. *Nat. Mater.* **2013**, *12*, 207–211.
- (42) Ramasubramaniam, A. *Phys. Rev. B* **2012**, *86*, 115409.